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BELL, Anthony <http://orcid.org/0000-0001-5038-5621> and HENDERSON, Michael

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# Comparison of cation-ordered $P2_1/c$ leucite structures with stoichiometry $K_2X^{2+}Si_5O_{12}$ (X = Mg, Fe, Co, Zn)

# <u>A.M.T.Bell</u><sup>1</sup>(Anthony.Bell@shu.ac.uk) and C.M.B.Henderson<sup>2</sup>.

1. MERI, Sheffield Hallam University, Sheffield, S1 1WB. 2. SEES, University of Manchester, Manchester, M13 9PL.

K2FeSi5O12 CMBH leucite;

# Introduction

Synthetic anhydrous analogues of the silicate framework minerals **leucite** (KAlSi<sub>2</sub>O<sub>6</sub>) and **pollucite** (CsAlSi<sub>2</sub>O<sub>6</sub>) can be prepared with the general formulae  $A_2BSi_5O_{12}$ and  $ACSi_2O_6$ , where A is a monovalent alkali metal  $\alpha$ cation, B is a divalent cation and C is a trivalent cation. These structures all have the same topology with B and cations partially substituting onto tetrahedrally coordinated sites (T-sites) in the silicate framework and charge balancing A cations sitting in extra-framework channels. The A cations can be replaced by ion exchange and these materials are of potential technological interest as storage media for radioactive Cs from nuclear waste [1]. We have used X-ray and neutron powder diffraction to determine and Rietveld [2] refine the crystal structures of many different leucite analogues [3-11], one of these was for  $K_2MgSi_5O_{12}$  [5], this was the first known leucite structure with ordered T-site cations and was determined from synchrotron X-ray powder diffraction data on the old Daresbury SRS [12, 13]. In this poster we report the Rietveld refinements of the crystal structures of three more cation ordered leucite analogues,  $K_2X^{2+}Si_5O_{12}$ where X = Fe, Co or Zn.



#### CMBH K2ZnSi5O12 Empyrean

() KZS1 fds.prf

700000

600000

500000

400000

![](_page_1_Figure_9.jpeg)

# **Synthesis**

The samples were made from appropriate stoichiometric mixtures of  $K_2CO_3$ ,  $SiO_2$ ,  $Fe_2O_3$  (X = Fe), CoO (X = Co) and ZnO (X = Zn). For X = Fe the sample was reduced from  $Fe^{3+}$  to  $Fe^{2+}$  [14].

![](_page_1_Figure_12.jpeg)

Data collection and analysis

All samples were mounted on low-background silicon wafers prior to ambient temperature X-ray powder diffraction data. Data were collected for the X = Fesample on a PANalytical X'Pert Pro MPD using Cu K $\alpha$  Xrays and an X'Celerator area detector. For the other two samples data were collected on a PANalytical Empyrean diffractometer with a PIXCEL-3D area detector. For X =Zn, Cu K $\alpha$  X-rays were used and for X = Co, Co K $\alpha$  Xrays were used. Analyses of the powder diffraction data showed that in all samples the major phase was isostructural with the  $P2_1/c$  structure of  $K_2MgSi_5O_{12}$  [5]. Rietveld refinements were done using FULLPROF [15], using the structure of  $K_2MgSi_5O_{12}$  as a starting model with X cations replacing Mg. However, for X = Co a second mica phase with the stoichiometry  $KCo_3(CoSi_3O_{10})OH_2$  (based on  $KCo_3(AISi_3O_{10})OH_2$  [16]) was included in the refinement as a 4.7(1) wt.% minor phase. For X = Fe and Zn single-phase refinements were done.

Figures 1, 2 and 3 show the Rietveld difference plots for these structures. Table 1 shows the comparison of some refined structural parameters for  $K_2MgSi_5O_{12}$  and X =Fe, Zn and Co. Ionic and crystal radii are from Shannon [17]. The tetrahedral angle variance [18] shows the distortion of tetrahedral units in the silicate framework. Figure 4 shows a VESTA structure [19] plot for

![](_page_1_Picture_16.jpeg)

cations, red spheres  $O^{2-}$  anions, blue tetrahedra represent  $SiO_4$  units and light brown tetrahedra represent  $FeO_4$ units.

# Table 1 - comparison of refined structural parameters

X	Mg	Fe	Zn	Со
a (Å)	13.168(5)	13.2574(5)	13.1773(2)	13.1878(1)
b (Å)	13.652(1)	13.6739(6)	13.6106(2)	13.6350(2)
c (Å)	13.072(5)	12.9240(5)	13.0248(2)	12.9876(1)
β (°)	91.69(5)	93.048(3)	91.70(6)	91.999(8)
V (Å <sup>3</sup> )	2348(2)	2339.6(2)	2334.98(6)	2333.96(4)
λ (Å)	1.52904	1.540598	1.540598	1.78901
instrument	SRS 8.3	X'Pert	Empyrean	Empyrean
ionic radius (Å)	0.71	0.77	0.74	0.72
crystal radius (Å)	0.57	0.63	0.60	0.58
mean <i>X</i> -O (Å)	1.90(1)	1.96(2)	1.89(1)	1.93(1)
mean K-O (Å)	3.349(6)	3.35(2)	3.364(6)	3.305(8)
O-Si-O variance	20.9(9.9)	183.1(130.9)	48.5(23.7)	65.9(33.1)
O-X-O variance	40.4(8.8)	230.3(125.1)	62.6(56.0)	75.3(82.6)
mean Si-O-Si (°)	140.9(10.7)	142.0(13.9)	144.7(12.6)	138.3(9.2)
mean Si-O-X (°)	130.6(9.8)	124.6(10.7)	128.1(12.5)	125.8(8.8)

### Discussion

Crystal structures have been refined for X =Fe, Zn and Co. These refined structures have complete T-site cation ordering and are isostructural with the  $P2_1/c$  structure of  $K_2MgSi_5O_{12}$ . Replacing Mg with a larger X cation causes a greater distortion of the SiO<sub>4</sub> and XO<sub>4</sub> tetrahedra the silicate framework. However, the refined structures with X cations also have a smaller unit cell volume than  $K_2MgSi_5O_{12}$ , the mean Si-O-X angles suggest a more collapsed framework for the structures with X cations compared to  $K_2MgSi_5O_{12}$ . It is interesting to compare the lattice parameter errors on data collected with

modern detectors to those determined from a 25 year old synchrotron X-ray powder diffraction dataset!

## Conclusions

Three new crystal structures for have been refined for the synthetic leucite analogues  $K_2 X^{2+}Si_5O_{12}$  where X = Fe, Co or Zn. These refined structures have complete T-site cation ordering and are isostructural with the  $P2_1/c$  structure of  $K_2MgSi_5O_{12}$ . However, replacing the  $Mg^{2+}$  cations in this structure with the larger Fe<sup>2+</sup>, Co<sup>2+</sup>, or Zn<sup>2+</sup> cations causes a greater distortion of the tetrahedral units and more collapsed silicate framework structures.

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