

**The relationship between local structure and photo-Fenton catalytic ability of glasses and glass-ceramics prepared from Japanese slag**

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**Table 1** Fe<sub>2</sub>O<sub>3</sub> (batch) content and bandgap energy of heat-treated slag samples calculated from DRS

Sample	Total Fe <sub>2</sub> O <sub>3</sub> content (mass % )	E <sub>g</sub> (eV)
SF11 Original	11.45 ± 0.01	2.15 ± 0.02
SF11 Melted	11.45 ± 0.01	2.09 ± 0.02
RSF14 Melted	14.40 ± 0.01	2.22 ± 0.04
SF30 Melted	30.00 ± 0.01	1.96 ± 0.01
SF50 Melted	50.00 ± 0.01	1.96 ± 0.01

**Table 2**  $^{57}\text{Fe}$ -Mössbauer parameters at room temperature of slag glass samples before and after heat treatment at 800 °C for 100 min

	Sample	assignment	A (%)	$\delta$ (mm s <sup>-1</sup> )	$\Delta$ (mm s <sup>-1</sup> )	$H_{\text{int}}$ (T)	$\Gamma$ (mm s <sup>-1</sup> )
Before	SF50 Melted	mag + mgh	32.2	0.30 $\pm$ 0.01	-0.02 $\pm$ 0.02	48.10 $\pm$ 0.01	0.42 $\pm$ 0.03
		mag.	34.4	0.52 $\pm$ 0.02	0.00	44.80 $\pm$ 0.02	0.88 $\pm$ 0.10
		Fe <sup>III</sup> $T_d$	23.6	0.27 $\pm$ 0.02	1.21 $\pm$ 0.03	-	0.84 $\pm$ 0.05
		hem.	9.8	0.37 $\pm$ 0.03	-0.20 $\pm$ 0.01	49.96 $\pm$ 0.02	0.43 $\pm$ 0.04
	SF30 Melted	Fe <sup>III</sup> $O_h$	100.0	0.40 $\pm$ 0.01	0.89 $\pm$ 0.02	-	0.99 $\pm$ 0.03
	RSF14 Melted	Fe <sup>III</sup> $T_d$	100.0	0.31 $\pm$ 0.01	1.25 $\pm$ 0.02	-	0.69 $\pm$ 0.02
	SF11 Melted	Fe <sup>III</sup> $T_d$	100.0	0.33 $\pm$ 0.01	1.25 $\pm$ 0.01	-	0.69 $\pm$ 0.02
	SF11 Original	Fe <sup>II</sup> $T_d$	85.5	1.00 $\pm$ 0.01	1.89 $\pm$ 0.02	-	0.63 $\pm$ 0.02
		Fe <sup>III</sup> $O_h$	14.5	0.50 $\pm$ 0.02	1.16 $\pm$ 0.08	-	0.51 $\pm$ 0.10
After	SF50 Melted	Fe <sup>III</sup> $T_d$	25.5	0.31 $\pm$ 0.03	1.14 $\pm$ 0.01	-	0.89 $\pm$ 0.01
		hem.	38.1	0.37 $\pm$ 0.01	-0.18 $\pm$ 0.04	51.40 $\pm$ 0.01	0.34 $\pm$ 0.03
		mgh.	36.4	0.37 $\pm$ 0.01	-0.02 $\pm$ 0.01	46.40 $\pm$ 0.01	1.23 $\pm$ 0.02
	SF30 Melted	Fe <sup>III</sup> $T_d$	100.0	0.32 $\pm$ 0.02	0.84 $\pm$ 0.03	-	0.76 $\pm$ 0.01
	RSF14 Melted	Fe <sup>III</sup> $T_d$	100.0	0.18 $\pm$ 0.01	1.34 $\pm$ 0.01	-	0.78 $\pm$ 0.01
	SF11 Melted	Fe <sup>III</sup> $O_h$	100.0	0.38 $\pm$ 0.04	0.92 $\pm$ 0.01	-	0.63 $\pm$ 0.01
	SF11 Original	Fe <sup>II</sup> $O_h$	9.8	1.10 $\pm$ 0.02	2.25 $\pm$ 0.05	-	0.44 $\pm$ 0.06
		Fe <sup>III</sup> $O_h$	90.2	0.37 $\pm$ 0.01	0.88 $\pm$ 0.01	-	0.75 $\pm$ 0.01

hem.: Hematite, mag.: Magnetite, mgh.: Maghemite,  $T_d$ : Tetrahedra,  $O_h$ : Octahedra, A: Absorption area,

$\delta$ : Isomer shift,  $\Delta$ : Quadrupole splitting,  $H_{\text{int}}$ : Internal magnetic field,  $\Gamma$ : Line width.

**Table 3**  $^{57}\text{Fe}$ -Mössbauer parameters at 77 K of slag samples heat-treated at 800 °C for 100 min

sample	assignment	A (%)	$\delta$ (mm s <sup>-1</sup> )	$\Delta$ (mm s <sup>-1</sup> )	$H_{\text{int}}$ (T)	$\Gamma$ (mm s <sup>-1</sup> )
SF50 Melted	Fe <sup>III</sup> $T_d$	14.1	0.40 $\pm$ 0.02	1.31 $\pm$ 0.03	-	0.92 $\pm$ 0.05
	hem.	49.0	0.48 $\pm$ 0.01	0.13 $\pm$ 0.01	53.50 $\pm$ 0.01	0.37 $\pm$ 0.01
	mgh.	36.9	0.43 $\pm$ 0.01	-0.11 $\pm$ 0.01	50.40 $\pm$ 0.08	0.71 $\pm$ 0.02
SF30 Melted	Fe <sup>III</sup> $T_d$	15.6	0.41 $\pm$ 0.02	1.50 $\pm$ 0.03	-	1.05 $\pm$ 0.05
	Mag + mgh	27.9	0.40 $\pm$ 0.03	-0.08 $\pm$ 0.05	41.90 $\pm$ 0.40	1.40 $\pm$ 0.10
	hem.	28.1	0.48 $\pm$ 0.01	0.01 $\pm$ 0.01	50.20 $\pm$ 0.10	0.65 $\pm$ 0.04
	mag.	28.4	0.42 $\pm$ 0.01	-0.04 $\pm$ 0.02	46.80 $\pm$ 0.10	0.73 $\pm$ 0.07
RSF14 Melted	Fe <sup>III</sup> $T_d$	58.5	0.29 $\pm$ 0.01	0.96 $\pm$ 0.01	-	0.46 $\pm$ 0.01
	Fe <sup>III</sup> $T_d$	41.5	0.24 $\pm$ 0.01	1.90 $\pm$ 0.01	-	0.40 $\pm$ 0.02
SF11 Melted	Fe <sup>III</sup> $O_h$	50.0	0.58 $\pm$ 0.01	1.09 $\pm$ 0.01	-	0.46 $\pm$ 0.03
	Fe <sup>III</sup> $T_d$	50.0	0.32 $\pm$ 0.02	1.21 $\pm$ 0.02	-	0.50 $\pm$ 0.03
SF11 Original	Fe <sup>III</sup> $T_d$	100.0	0.42 $\pm$ 0.01	1.20 $\pm$ 0.01	-	1.03 $\pm$ 0.03

hem.: Hematite, mag.: Magnetite, mgh.: Maghemite,  $T_d$ : Tetrahedra,  $O_h$ : Octahedra, A: Absorption area,

$\delta$ : Isomer shift,  $\Delta$ : Quadrupole splitting,  $H_{\text{int}}$ : Internal magnetic field,  $\Gamma$ : Line width.