

Tetrasubstituted copper phthalocyanines : correlation between liquid crystalline properties, films alignment and sensing propertie

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IX.	(1)	(2*)	(3*)	(4)	Route
-OC ₈ H ₁₇	1a	2a	3a	4a	i, ii, iii
-O ₁₆ H ₃₃	1b	2b	3b	4b	i, ii, iii
-SC8H17	1c	2c	3c	4c	i, ii, iii
-SC ₁₆ H ₃₃	1d	2d	3d	4d	i, ii, iii
-O(CH ₂ CH ₂ O) ₃ CH ₃	1e			4e	iv
-S(CH ₂ CH ₂ O) ₃ CH ₃	1f			4f	iv

Figure 1. Synthesis of CuPc derivatives, **4a-4c**, **4d** [34], **4e-f** [35]. *i*: anhydrous PbO, solventfree, 210°C, 5h. *ii*: CH₃COOH. *iii*: anhydrous CuCl₂, anhydrous n-hexanol, reflux, 2h. *iv*: anhydrous CuCl₂, anhydrous n-hexanol, DBU, reflux, 24h. *PbPc (**2**) and H₂Pc (**3**) derivatives were not purified but directly used for the synthesis of CuPc derivatives (**4a-d**).



Figure 2. POM images of CuPcs (4a-f). The scale bar indicates 100 $\mu m.$



Figure 3. XRD patterns of CuPc derivatives (4a-f) measured at room temperature.



Figure 4. Optical absorption spectra of CuPc **(4a-f)**: solutions in dichloromethane (a), asdeposited films (b), films after heat treatment (c).



Figure 5. AFM images of the films of CuPcs (4a-f).



Figure 6. Polarized Raman spectra of **4b** (a) and **4d** (b) films on glass substrates after heat treatment.



Figure 7. Response curves of the film of **4c** to ammonia vapours at concentrations of 10-50 ppm (a) measured at 50%. Response curves of the films of **4c** (b) and **4f** (c) to ammonia vapours (30 ppm) measured at different relative humidities.



Figure 8. Normalized sensor response vs NH₃ concentration, measured at RH 50%.



Figure 9. Normalized sensor response of 4c and 4f films vs NH₃ concentration, measured at RH 0, 50 and 75%.